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Effects from high-speed civil traffic aircraft emissions on polar stratospheric clouds.

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This report constitutes the DMI-contribution to the final report of the EU project "Modelling of the impact on ozone and other chemical compounds in the atmosphere from airplane emissions" (AEROCHEM II, contract ENV4-CT97-0621, 1998-2000), funded by the European Commission Environment and Climate programme (4th Framework programme).

1. Introduction.

Emissions from stratospheric high-speed civil traffic (HSCT) aircraft may influence the probability of formation and the properties of polar stratospheric clouds (PSC). PSC particles play a decisive role for chemical ozone depletion in polar regions. The influence of PSC particles is two-fold. First, on the surfaces of the cloud particles, heterogeneous chemical reactions convert reservoir halogen compounds into potentially ozone destroying radicals, secondly the cloud particles may irreversibly remove reactive nitrogen (and water vapour) by gravitational sedimentation which may prolong the ozone depletion [*WMO*, 1999].

The cloud particles are composed of nitric acid and water, and stratospheric sulphate aerosols are believed to constitute the background particles, which serve as sites for PSC particle growth at low temperatures. New sulphate aerosol particles could be produced as a result of aircraft plume processing of SO_2/SO_3 and enhanced concentrations of SO_2 , eventually oxidising into H_2SO_4 , may in general increase the stratospheric sulphate mass mixing ratios and background aerosol concentrations. More directly may aircraft emissions of water vapour and NO_x , converting into H_NO_3 , lead to higher threshold temperatures for PSC formation and enhanced PSC particle surface area densities.

Only a few studies, applying more detailed PSC-microphysical schemes, have been performed to study the effects of stratospheric aircraft emissions on PSC properties [Peter et al., 1991; Pitari et al., 1993; Considine et al., 2000]. A thorough review of the effects on PSC surface area densities, applying a selection of state-of-the-art atmospheric models with varying degrees of microphysical complexity, has been provided by IPCC [1999, cp. 4]. Common to most studies is the assumption of the existence of two types of solid phase PSC particles, forming at certain threshold temperatures (e.g. the nitric acid trihydrate (NAT) condensation temperature and the ice frost point). Subsequent research has revealed the existence of at least three PSC particle types, showing complicated transition processes between liquid and solid type particles which depend on the temperature histories of individual particles [e.g. Larsen, 2000a]. In this study a different approach has been taken, applying domain filling trajectories of a given Arctic winter, coupled with a detailed microphysical model, to investigate the effects of aircraft-enhanced concentrations of nitric acid and water vapour on PSC formation. The aircraft induced perturbations to HNO₃ and H₂O have been calculated by two 3D chemical transport models (3D-CTM), applying emissions scenarios provided by NASA. Although the threshold temperatures for the existence of different types PSC particles are marginally changed by increased concentrations of H₂O and HNO₃, the detailed microphysical simulations as presented below reveal a substantial enhancement of solid type PSC formation and the potential for stronger denitrification and dehydration.

2. PSC microphysics.

Two types of polar stratospheric clouds have been observed above the ice frost point: liquid particles (type 1b PSC), composed of supercooled ternary solutions (STS; $H_2SO_4/HNO_3/H_2O$), and solid particles (type 1a PSC), presumably composed of nitric acid trihydrate which is the stable nitric acid hydrate at stratospheric conditions. Furthermore, larger type 2 ice PSC particles form below the ice frost point temperature. The heterogeneous chemical reactions taking place on the surfaces of the PSC particles, activating halogen species, depend on the chemical compositions and physical phase of the particles. In addition, only solid type PSC particles are

responsible for denitrification and dehydration, i.e. the irreversible removal of reactive nitrogen and water by gravitational sedimentation of nitric acid holding particles.

The microphysical model from the Danish Meteorological Institute [Larsen, 2000b] comprehends all these PSC particle types. The model calculates the time dependent polar stratospheric cloud particle size distributions and chemical compositions together with changes in gas phase mixing ratios of water vapour and nitric acid vapour, assuming an initial size distribution of background sulphate aerosols and initial concentrations of H₂O and HNO₃. The model simulates the formation, growth, evaporation, and sedimentation of type 1b PSC particles (supercooled ternary solution, STS), type 1a PSC particles (assumed to be composed of nitric acid trihydrate, NAT), type 2 PSC ice particles, and frozen sulphate aerosol particles, assumed to be composed of sulphuric acid tetrahydrate (SAT). The model applies the basic vapour diffusion equation to calculate the exchange of mass between the gas and condensed phase during particle growth and evaporation. The model comprehends a number of possible pathways for phase changes and the formation of solid type PSC particles. This includes the calculation of homogeneous freezing rates of ice in STS particles to form solid type 2 PSC particles a few K below the ice frost point. The model also allows for heterogeneous nucleation of pre-activated SAT for the formation of type 1a PSC NAT particles in addition to deliquescence of SAT particles during cooling below the STS threshold temperatures and SAT melting at higher stratospheric temperatures. The model is written as an atmospheric box model, facilitating a coupling to chemical transport models and photochemical trajectory models.

One major obstacle to prescribing PSC formation in larger atmospheric chemistry models is the inherent temperature hysteresis in the particles' life cycle, changing between the liquid and solid phases. At decreasing temperatures the background sulphate aerosol particles take up nitric acid and water vapour, continuously turning into fully developed type 1b PSC particles at roughly 3-4 K below the existence temperature of NAT (T_{NAT} , roughly 195 K). The growth and changing in composition of type 1b PSC particles is illustrated in Figure 1. These particles survive in the liquid state, even to temperatures a few degrees below the ice frost point (8-10 K below T_{NAT}). The particles eventually freeze into type 2 PSC ice particles. During subsequent heating a residual of type 1a PSC particles can exist in the solid state at all temperatures below T_{NAT} before the remnant, possibly a solid sulphuric acid particle, melts at 210-220 K after all nitric acid in the particle has evaporated above T_{NAT} . Pathways may also exist for dissolution of solid sulphate aerosols, turning into type 1b PSC at low temperatures. NAT may also nucleate on pre-activated SAT particles at decreasing temperatures. Thereby the occurrence of a specific PSC particle type will depend on the actual temperature history and can only be modelled more accurately by a microphysical model when individual air parcel trajectories are known.

Figure 1 also illustrates the sensitivity of calculated type 1b PSC particle volumes and threshold temperatures to changes in nitric acid and water vapour gas phase concentrations. In these calculations the H_2O and HNO_3 gas phase concentrations have been changed by 20%. In particular the water vapour concentrations determine the ice frost point temperature and the homogeneous freezing rates of ice in STS. Increasing water vapour concentrations (e.g. caused by aircraft emissions) will imply higher freezing temperatures and more frequent solid type PSC formation and thereby the potential for enhanced denitrification. In focus of this project has been a study of expected changes in PSC properties due to changes in concentrations of H_2O and HNO_3 induced by stratospheric supersonic air traffic.



Figure 1. Type 1b PSC (STS) particle equilibrium chemical composition (upper panel) and volume (lower panel), calculated as function of temperature [*Carslaw et al.*, 1995] at 35 hPa, assuming 5 ppmv H₂O, 10 ppbv HNO₃, and 0.4 ppbv H₂SO₄ (black curve). The upper panel shows how the chemical composition continuously changes from a nearly binary H₂SO₄ solution through a ternary composition into a nearly binary HNO₃ solution at decreasing temperatures. In the lower panel, the blue curves indicate the volume range assuming 5±1 ppmv H₂O and the green curves assuming 10±2 ppbv HNO₃. The two red curves indicate the particle volumes, assuming 10 and 100 times increase in H₂SO₄, typical for volcanic conditions. T_{NAT} and T_{ice} are indicated on the temperature axis with the H₂O-variation range (blue square symbols) and the HNO₃-variation range (green circle symbols). The temperature axes in the lower part indicate the temperature existence range of different stratospheric particle types, showing the temperature hysteresis with freezing of liquid particles 3-4 K below T_{ice} and melting around 216 K.



temperature surface 475 K on 16 January 1990 at 12.00 UT. From this time in mid-winter, the trajectories are calculated backward in time until 1 December 1989 and forward in time until 1 March 1990 with a time resolution of 2 hours. The same holds for the other 8 model levels, giving a total of 23300 trajectories. The vortex edge is defined at 36 PV units at this potential temperature and at an equivalent modified PV value at other levels. (1 PVU = 10^{-6} K m²s⁻¹kg⁻¹).

3. Domain filling trajectories.

Many previous studies [*IPCC*, 1999] have calculated PSC properties in fixed grid points in larger atmospheric circulation models. In this investigation, calculations of the occurrence and surface area densities of PSCs are performed in a Lagrangian way, based on coupled domain filling trajectories and detailed microphysical simulations and using analysed temperatures from a representative winter (1989/1990) in the northern hemisphere. Thereby the temperature histories of individual PSC particles are used as input to calculate PSC properties. The winter 1989/1990 was relatively cold from mid December to mid February when the area with temperatures below T_{NAT} at 30 hPa covered between 2 and 4% of the NH (25-50 % of the area north of 67°N) [*Pawson et al.*, 1995; *Manney et al.*, 1994]. The same representative temperatures from this winter have been used in different scenarios, assuming aircraft induced perturbations to the gas phase concentrations of H₂O and HNO₃.



The domain-filling trajectory calculations use 6 hourly ECMWF (European Centre for Medium-Range Weather Forecasts) analyses in a 1.5 degree latitude-longitude grid. They are performed in an equal-area grid with a grid-distance of 139 km (1.25° of latitude) and start at 9 isentropic levels inside or at the edge of the polar vortex and north of 60°N. The trajectory data are stored for every 2 hours. The trajectories are calculated both backward and forward in time for 45 days from 16 January 1990 covering the whole PSC season in this winter from the beginning of December 1989 to the end of February 1990. The results are 23300 trajectories of 90-day duration, which are necessary for the PSC box model, but which become increasingly inaccurate away from 16 January 1990. The bulk advection is, however, reasonably well represented in such extended calculations [e.g. *Manney et al.*, 1995; *Morris et al.*, 1995; *Knudsen and Grooβ*, 2000]. The potential temperature in each time step is decreased by the diabatic heating in each time step. The clear-sky heating rates are calculated with the ECMWF radiation scheme [*Morcrette*, 1991] and they are described in *Knudsen and Grooβ* [2000].

The location of the initial points on model level 5 at 475 K potential temperature is illustrated in Figure 2 showing the temperature and potential vorticity. Since most of the trajectory points are initialised inside the polar vortex there is relatively little geographical spread in the trajectories going backward and forward in time from the mid-winter initial time. In the following calculations the vortex edge has been set at a potential vorticity of 36 PV units at potential temperature θ =475 K and at the equivalent value of modified potential vorticity [MPV=PV·(475K/ θ)^{4.5}] at the other potential temperature surfaces. Radiative cooling leads to an average decrease in potential temperature throughout the winter. Figure 3 illustrates the average potential temperature and standard deviation for all trajectories on the 9 model levels.

In Figure 4 is shown the fraction of locations on the nine model levels inside the polar vortex with temperatures below T_{NAT} . Temperatures reach the lowest values with most PSC formation on levels 4, 5, and 6 and calculation on these levels will be in focus in the following analysis.

The temperature histories from the trajectories have been used as input to the microphysical model to calculate PSC particle surface chemical areas, compositions, physical phase of the particles, and gas phase concentrations of HNO_3 and H_2O at each trajectory point throughout the winter.

The microphysical model has been initialised with aerosol surface areas, consistent with a SAGE I and II surface area climatology [*Hitchman et al.*, 1994]. Figure 5 shows



Figure 5. Applied sulphate aerosol surface area densities, consistent with SAGE extinction measurements [*Hitchman et al.*, 1994]

the applied surface area densities. Number densities of sulphate aerosols (N) are derived from the surface area densities (A), assuming that the particles are characterised by a lognormal size distribution with median radius $r_m=0.0725 \ \mu m$ and a geometric standard deviation $\sigma=1.86$, i.e. $A = N 4 p r_m^2 \exp[2(\ln s)^2]$. Initial concentrations of nitric acid and water vapour are provided by the 3-D CTM at University of Oslo and the University of 1'Aquila, applying different aircraft emission scenarios.

Figure 6 illustrates an example of calculated fields of type 1a and type 1b PSC surface area density and gas phase concentrations of HNO_3 , based on background ("no-aircraft") HNO_3 and H_2O conditions from the Oslo CTM model.



Figure 6 Calculated surface area densities of type 1b PSC (upper left panel) and type 1a PSC (upper right) together with gas phase HNO_3 (lower right) on 7 February 1990, 12.00 UT at model level 6. The lower left panel shows the temperature field.

4. Chemical transport models.

Both the stratospheric chemical transport models at the University of Oslo (UiO) and at the University of l'Aquila (ULAQ) have been used to provide perturbed fields in HNO_3 and H_2O due to aircraft emissions of nitrogen oxides and water vapour.

The SCTM-1 at UiO is a global 3-dimensional stratospheric chemical transport model. The resolution is 7.8° latitude x 10° longitude in 29 layers extending from the surface up to 90 km. The vertical layers are defined in sigma coordinates below 100 hPa and in pressure coordinates above. The transport is driven by wind fields from the GISS GCM [*Rind et al.*, 1988], while temperatures are from daily National Center for Environmental Prediction (NCEP) global analyses for 1990.

The chemistry code [*Stordal et al.*, 1985, *Isaksen et al.*, 1990] includes 55 species and focuses on stratospheric chemistry. 104 thermal and 47 photolytic reactions are integrated by the QSSA method [*Hesstvedt et al.*, 1978]. Although the chemistry module has not been designed for detailed tropospheric or mesospheric studies, ozone production from methane is included so that the code is basically applicable in the upper troposphere as well. Seven heterogeneous reactions are simulated on stratospheric sulphate aerosols and/or on type 1 PSCs including heterogeneous processing of reactive chlorine, bromine, and nitrogen. The photolysis rate calculation is done on-line by the module of *Kylling et al.* [1995], recalculating and archiving the diurnal cycles for the 3-D model grid every 7 days. The ozone distribution, air density, and temperatures used in this calculation are taken directly from the model during runtime. As the time step for chemistry integrations is 10 minutes and the photolysis rates are resolved at intervals of 40 min, the chemical solution is iterative.

The advection scheme applies the concept of Second Order Moments [*Prather*, 1986], which is an upstream scheme taking into account the sub-grid scale distribution of mass and transported tracers. The use of the moments leads to an effective resolution that is 3 times the formal grid resolution. Tropospheric convection and horizontal diffusion in SCTM follow *Prather et al.* [1987,1990]. In moist convection, 50% of the mass is taken from the originating layer and put to the receiving layer, while dry convecton causes complete mixing. While the transport algorithms cover the whole model domain, chemistry integrations are done only between about 6 and 55 km. Boundary conditions at the surface and the model top are derived from multi-year runs of the OSLO 2D model [*Isaksen et al.*, 1990]. Mesospheric initialization of SCTM is done by calculating scale heights at 50 km, which are used to extrapolate the fields from the Oslo 2D model upward. For a detailed description of the original 21-layer version see *Rummukainen* [1996] and *Rummukainen et. al.* [1999]. It must be noted that in this study a higher resolution was implemented in the upper troposphere and the lower stratosphere, taking advantage of the SOM transport scheme, whereby the number of model layers increased to 29.

As a model year, 2015 was chosen to illustrate future aircraft impact. The background atmosphere is based on the IPCC emission scenario IS92a [*IPCC*, 1995]. The scenarios for aircraft emissions are from the NASA data base [*Baughcum et. al*, 1998]. Emissions of NO_x and water vapour from sub- and supersonic aircraft were considered. Three different cases were calculated: 1) 'No aircraft' emissions, 2) subsonic emissions only, and 3) both sub- and supersonic emissions (designated 'supersonic' scenario in the following). Each case was integrated for 7 years with constant emissions in order to get stable results for the perturbations.

For the supersonic aircraft emissions a fleet consisting of 500 aircraft with an emission index E.I.(NOx)=5 was chosen (5 grams of NO_x are emitted for each kg of burnt fuel). For H₂O emissions the assumption was E.I.(H₂O)=1230. The perturbations were calculated in a 3-dimensional grid for 01 December 2015 for each of the three cases. Modelled HNO₃ is increased by up to 25 % at mid northern latitudes due to the inclusion of subsonic aircraft.

Supersonic aircraft lead to a secondary maximum increase of 12% in the lower tropical stratosphere. Subsonic aircraft do not perturb water vapour concentrations significantly, while an increase of up to 14% due to supersonic aircraft is modelled at mid northern latitudes at about 18 km altitude.

Figure 7 illustrates background ('no aircraft') concentrations of H_2O and HNO_3 and the calculated perturbations in H_2O and HNO_3 for the 'supersonic' scenario in the UiO model.





As a rough estimate, the ice frost point temperature will increase by a little more that 1 K for an increase in water vapour of 1 ppmv, meaning that the H_2O perturbations shown in Figure 7 will imply an increase in frost point temperatures on the order of 0.2-0.4 K. An increase of HNO₃ by 1 ppbv only causes an increase in the condensation temperature of NAT (T_{NAT}) of a few tenths of a degree whereas T_{NAT} , like the frost point temperature, increases by roughly 1K for an increase of 1 ppmv H_2O . Therefore the perturbations in H_2O will have the strongest influence on the aircraft induced perturbations.

In the University of l'Aquila model, all calculations were made with a low-resolution 3D-CTM. The basic features of the model are as follows. The grid is made by 26 log-pressure levels extending from the surface up to about 71 km altitude, while horizontally there is a 10x20 degrees resolution in latitude-longitude. The chemical code includes stratospheric families and the most important hydrocarbons for a realistic tropospheric chemistry parameterisation. Transport includes advection, vertical diffusion, weak horizontal diffusion and deep convection. A microphysical code for SSA, PSC (NAT and ice), and tropospheric aerosols (carbonaceous, dust, seasalt) is on-line with the model in a gas-particle mass conserving framework.

Subsonic and future supersonic aircraft are included using the NASA emission scenarios released in 1996 and appropriate for the year 2015 (data available on the Langley UADP database). The corresponding version of the 2D model has used the same emission data for the *IPCC* [1999] assessment. HSCT emissions considered here are those of NO_x , H₂O, SO₂, black carbon soot, CO, CO₂ and hydrocarbons (as CH₄). Additional formation of sulphate particles is obtained in two ways: enhanced condensation of gas phase H₂SO₄ (formed through SO₂ oxidation) and direct particle production in aircraft plumes, assuming a 10% conversion fraction of the emitted SO₂. Emission indices used here and those realistically predicted for the year 2015 technology, that is (in g/kg-fuel): 1237 (H₂O), 5 (NO_x), 0.4 (SO₂), 0.04 (soot), 1.0 (CO), 3155 (CO₂), 0.2 (CH₄).

 H_2O emissions may also perturb the radiation budget in the lower stratosphere, with a potential feedback on species transport; this effect has not been deeply investigated so far. The ULAQ group has made some preliminary studies using a stratospheric GCM.

Water vapour heats the lower stratosphere by absorbing black body radiation from the troposphere and near infrared radiation from the sun, and cools the stratosphere by emitting longwave radiation to space. The combination of heating and cooling produces net heating rates that are positive in the lower stratosphere tropical region and negative at mid-high latitudes, mainly in the winter hemisphere. This is because the tropical lower stratosphere is colder (less cooling to space), the underlying troposphere is warmer (more heating form below), and because there is less incoming solar radiation in the winter mid-high latitudes.

The contribution of H_2O to the stratospheric diabatic circulation will add to those of O_3 and CO_2 : the Brewer-Dobson circulation shows on average tropical upwelling and wintertime extratropical subsidence. With an increase of stratospheric H_2O , like that produced by supersonic aircraft, we may expect an enhancement of the diabatic circulation, with a resulting perturbation of the stratosphere-troposphere ozone flux as well as changes in the large-scale distribution of other chemical tracers in the lower stratosphere (NO_y , Cl_y , H_2O itself). Concentrations of H_2O and HNO_3 , calculated by the ULAQ model and including the H_2O radiative feedback, will be designated the 'supersonic-radiative' scenario in the following. Figure 8 shows the background concentrations of H_2O and HNO_3 , and the perturbations in the 'supersonic' and 'supersonicradiative' scenarios.

Calculations have been performed with the microphysical model, using the same trajectories and temperature histories as for the 1989/1990 winter, but assuming perturbed fields nitric acid and water vapour, based on the 2015 scenarios of NO_x and H_2O emissions from subsonic+supersonic aircraft, calculated by the CTMs at the University of Oslo and University of l'Aquila. Calculations have also been performed for the subsonic aircraft scenario, showing only a modest influence on the PSC properties. Therefore, only the 'supersonic' scenarios will be compared to the 'no aircraft' emission scenario in the following.



row) perturbations.

5. Calculated PSC properties.

Figure 9 shows changes in occurrence of locations inside the polar vortex with temperatures below T_{NAT} in model levels 4, 5, and 6. The upper curves in each panel show the results for the UiO 'no aircraft' scenario (similar to Figure 4) whereas the thin lower curves show the difference between 'supersonic' and 'no aircraft' conditions. The fraction of locations inside the polar vortex with temperatures below T_{NAT} typically increase 2-5% and up to 10% in model level 4 in early winter due to the changes in HNO₃ and H₂O.



The type 1b PSC equilibrium model of *Carslaw et al.* [1995] has also been used to calculate the associated changes in vortex-averaged particle volumes (cf. Figure 1) due to changes is HNO_3 and H_2O concentrations, calculated by the UiO model as shown in Figure 10. The increased gas phase concentrations give rise to increases in type 1b PSC particle volumes of about 10% inside the polar vortex in the 'supersonic' conditions, compared to the 'no aircraft' scenario.



Figure 10. The STS type 1b PSC equilibrium model of *Carslaw et al.* [1995] has been used together with temperatures from the trajectory model to calculate average particles volumes inside the polar vortex at model levels 4, 5, and 6. The upper thick curve shows the vortex-averaged type 1b PSC particle volumes for the 'no aircraft' conditions (UiO model) and the lower thin curve the difference in calculated particle volumes between the 'supersonic' and ' no aircraft' scenarios.

The calculations of changes in occurrence of temperatures below T_{NAT} (Figure 9) and volumes of type 1b PSC particles (Figure 10) are only threshold conditions, based exclusively on vapour pressures of NAT and STS, respectively, without any microphysical assumptions about the actual PSC formation under the changing conditions. Therefore the temperature histories from the trajectories have been used as input to the microphysical model, performing simulations of the PSC formation under the two aircraft emission scenarios. Figure 11 shows the calculated frequency of occurrence of three different types of PSCs inside the polar vortex. The frequencies

are calculated as the percentages of trajectory points inside the polar vortex where the model shows that PSC particles of the different types exist. The gray shaded curves show the 'no aircraft' scenario and the thick curves the 'supersonic' scenario, using the University of Oslo model perturbations.

First it should be noticed from the 'no aircraft' results that the frequency of PSC occurrence is substantially smaller that the frequency when temperatures drop below T_{NAT} (cf. Figure 9), pointing out that assumptions about PSC formation below T_{NAT} could be a simplified assumption to be applied in atmospheric chemistry models, in particular in marginal cold winters.

As it would be expected the 'supersonic' scenario emission only give rise to an increase in type 1b PSC occurrence of a few percent since the formation of this type of clouds only depend on the STS threshold temperature conditions.

However, a much more pronounced difference between the two scenarios is seen in the occurrence of solid type 1a and type 2 PSCs. Solid type PSCs are assumed only to form when ice freezes out of STS solution 3-4 K below the ice frost point. The homogeneous freezing rate depends on the water vapour partial pressure (and temperature) and increased concentrations of H_2O lead to more widespread freezing and solid type PSC formation.

Figure 12 shows the vortex-averaged surface area densities of the different PSC types. A substantial increase in solid type PSC surface area is seen between the two emission scenarios. It should be noticed that in particular the increase in type 2 PSC ice particle surface areas may have implications for heterogeneous chemical activation.

It is also important to note that only solid type PSC particles may induce denitrification and dehydration. A potential exists that the enhanced H₂O concentrations, caused by stratospheric aircraft emissions, may lead to more solid type PSC formation and thereby more widespread denitrification.

In order to quantify this potential for enhanced denitrification, model runs have also been performed, using the University of Oslo model perturbations, where the gravitational sedimentation of particles is included. In principle it is not possible in a domain filling trajectory approach to make an accurate calculation of denitrification over a large span in altitude since the in-fall of particles from layers above (renitrification) cannot be exactly represented. However, the fall-out of particles from trajectories in a given layer can be calculated with high accuracy and thereby how much HNO_3 and H_2O is left in the gas phase after all remaining particles are evaporated. Results from a calculation of this kind are shown in Figure 13.



 H_2O and HNO_3 concentrations.





The difference between the black and red curves in Figure 13, representing the 'no aircraft' and 'supersonic' scenarios and not including particle sedimentation simply reflects the enhanced mixing ratios caused by the aircraft emissions. The temporary drops in mixing ratios reflect the uptake from the gas phase by condensation into the particles.

The results from the calculations including sedimentation show the difference in mixing ratios between the 'supersonic' and 'no aircraft' mixing ratios decreases throughout the winter, meaning that a larger denitrification and dehydration takes place in the 'supersonic' scenario. At model level 5, where the relative increase in occurrence of solid type PSC is largest, the HNO₃ mixing ratio decrease from 4.47 to 3.89 ppbv due to particle fall-out in the 'no aircraft' scenario (green curve in left-hand panel in Figure 13). The decrease is from 4.76 to 3.93 ppbv HNO₃ in the 'supersonic' scenario (blue curve). In the no-sedimentation cases there is a slight drop in HNO₃ mixing ratios of 0.08 ppbv (due to transport). Correcting for this means that there is a potential for stronger denitrification of about 38%. Likewise the 'no aircraft' scenario shows a drop in H₂O mixing ratio of about 0.24 ppmv from 3.92 to 3.68 ppmv (green curve in right-hand panel), compared to a drop of 0.43 ppmv from 4.29 to 3.86 ppmv in the 'supersonic' case, i.e. an increase in potential dehydration of 68% after correcting for the transport effects of 0.04 ppmv.



Calculations have also been performed for model level 5, using the H_2O and HNO_3 concentrations and aircraft induced perturbations from the University of l'Aquila model. Figure 14 shows the calculated frequency of occurrence of the three PSC particle types and Figure 15 the calculated vortex-averaged surface area densities.

First it can be noticed that the frequency of occurrence in the 'no aircraft' scenario, using the University of l'Auila data compared to the Oslo data, are larger due to higher HNO_3 concentrations (cf. Figure 11). Also the frequency of solid type PSC is higher due to higher H_2O concentrations (cf. Figures 7 and 8).



Again, a pronounced difference in the occurrence and surface area densities for solid type PSC particles is calculated, caused by the "supersonic" aircraft emissions whereas the influence on liquid type 1b PSC is moderate. The "supersonic-radiative" scenario induces nearly no effects on the type 1b and type 2 PSC properties, compared to the "supersonic" case, and only a smaller increases in occurrence and surface areas of type 1a PSCs (compare upper and lower rows in Figures 14 and 15).

6. Conclusions.

Microphysical simulations using domain filling trajectories have been performed to investigate the influence on PSC properties of enhanced concentrations of HNO₃ and H₂O, caused by aircraft emissions in year 2015 scenarios. The results show that emissions from a projected fleet subsonic aircraft only cause minor influences on PSC occurrence and surface area densities. However, the effects from a combined fleet of subsonic and supersonic aircraft clearly show up in the simulations. Increased concentrations of HNO₃ and H₂O in the "supersonic" scenarios on the order of 5-10 % will have a direct influence on the thermodynamic threshold temperatures for the existence of NAT, STS, and ice in the NH polar stratosphere. This may cause the frequency of occurrence of temperatures below these thresholds to increase by 2-5% and give rise to increases in Type 1b PSC equilibrium volumes of about 10 %.

Performing more detailed microphysical simulations, which allow the inherent temperature hysteresis in the life cycles of different PSC particle types to be taken into account, using temperature histories from a large number of individual airparcels, the influences of the supersonic aircraft emissions become more pronounced. In particular, the increased concentrations of water vapour imply higher freezing temperatures for ice in STS Type 1b PSC particles, required for the formation of solid Type 1a and Type 2 PSCs. Both the frequencies of occurrence and the vortex-averaged surface area densities of solid type PSCs may increase by 50-100% in continuous periods throughout the winter. Although the surface areas of these types of PSCs are relatively small, compared to the much more abundant liquid Type 1b PSCs, only the solid type PSC could be responsible for significant denitrification and dehydration of the Arctic stratosphere. Detailed multi-level sedimentation calculations are not possible within a domain-filling trajectory approach. However, detailed calculations of the gravitational particlefall-out of individual airparcels and model-level averages of exchange rates in HNO₃ and H₂O between layers can be made. These calculations show that the "supersonic" aircraft emissioninduced formation of solid type PSC particles may increase the denitrification rate by about 38 % and even lead to larger relative increases in dehydration. It should be noticed that these conclusions depend on the winter chosen (1989/90) which was relatively cold.

Climate model calculations [*Shindell et al.*, 1998] have predicted that NH stratospheric December-January temperatures may decrease by 3-10 K in 2010-2019 due to increased concentrations of greenhouse gases, less frequent stratospheric warmings, and ozone depletion. For a rough comparison between possible "supersonic" aircraft effects and decreased stratospheric temperatures, the calculations on model level 5, using the University of Oslo "no-aircraft" concentrations of HNO₃ and H₂O, have been repeated with all temperatures lowered by 3 K. The results are shown in Figure 16. Quite obviously, the formation of solid type PSC is much more sensitive to this change in temperatures conditions, compared to the aircraft perturbations of H₂O, which only increase the frost point temperatures on the order of 0.2-0.4 K. However, both the possible decreasing stratospheric temperatures, due to supersonic aircraft emissions, work in the same direction, mainly affecting the solid type PSC formation, which may lead to increased denitrification and possible dehydration



Figure 16 The upper row shows the frequency of occurrence of different types of PSC particles at model level 5, calculated using the University of Oslo 'no aircraft' concentrations of HNO_3 and H_2O (hatched curves, cf. Figure 11), compared to a 'no aircraft' simulation with all temperatures lowered by 3 K (thick curves). The lower row shows the calculated vortex-averaged surface area densities in the same two simulations (cf. Figure 12).

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